# Indolizines, Triazolo[4,3-a]pyridines, Benzimidazo[1,2-d]oxadiazoles, and Pyrazolo[1,5-c]triazoles via Nitrogen and Sulfur Ylides

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ABSTRACT: The pyridinium salts **2a,b** reacted with dimethyl acetylenedicarboxylate (DMAD) to give the indolizine derivatives **6a,b**. Pyridinium salts **2a,b** also reacted with pyrazole-5-diazonium salt to afford the hydrazonoyl bromides 8a,b, which on treatment with aqueous ethanolic sodium carbonate furnished the 8aH-1,2,4-triazolo [4,3-a] pyridine 10. When sulfonium bromide 11 was treated with nitrous acid and with pyrazole-5-diazonium salt, it afforded the new hydroximoyl and hydrazonoyl halides 12 and 17, respectively. Compound 12 reacted with 2-methylthiobenzimidazole to furnish benzimidazo-[1,2-d]-1,2,4-oxadiazole derivative 14. Treatment of either 12 with 3-phenyl-5-aminopyrazole or 17 with triethylamine resulted in the formation of the same product: pyrazolo[1,5-c]-1,2,4-triazole derivative 16. © 2004 Wiley Periodicals, Inc. Heteroatom Chem 15:432-436, 2004; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.20037

#### INTRODUCTION

Indolizine derivatives are potential pharmaceutical candidates [1–3] and are useful as photographic sen-

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sitizers [4,5]. In addition, benzimidazole and benzothiazole derivatives were found to be effective anti-HIV and anticancer agents [6–8]. As part of our research interest toward developing new routes for the synthesis of fused heterocyclic systems having benzimidazole and benzothiazole moeities [9–12], we establish here a novel route to the incorporation of such moieties into the entitled bridged-head nitrogen heterocycles utilizing some new nitrogen and sulfur ylides.

## RESULTS AND DISCUSSION

Thus, pyridinium bromide 2a, which was recently reported by us [10], was treated with dimethyl acetylenedicarboxylate (DMAD) as dipolarophile in dry benzene at refluxing temperature, in the presence of triethylamine, and resulted in the formation of a single yellow-colored product as examined by thin-layer chromatography (TLC). Elemental analysis and mass spectra established the molecular formula of the product as C<sub>20</sub>H<sub>14</sub>N<sub>2</sub>O<sub>5</sub>S. Spectral data (IR, <sup>1</sup>H and <sup>13</sup>C NMR) were in complete agreement with the assigned indolizine structure **6a**, Scheme 1. The <sup>1</sup>H NMR spectrum of compound **6a** revealed two singlet signals at  $\delta$  3.56 and 3.85 due to two methyl ester groups in addition to the aromatic multiplets in the region  $\delta$  7.36–9.54. Its IR spectrum also showed two characteristic carbonyl absorptions at 1732 and 1695 cm<sup>-1</sup>. In the same manner, pyridinium bromide 2b reacted with DMAD under the same

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#### **SCHEME 1**

reaction conditions to afford the indolizine derivative **6b**, Scheme 1. Indolizine derivatives **6a,b** are assumed to be formed via 1,3-dipolar cycloaddition of DMAD to the nitrogen ylides 4a,b (which were formed in situ) to give the nonisolable intermediates **5a,b**, which are oxidized under the reaction conditions to give the indolizine products **6a,b**. This reaction proceeded in a similar way as the one reported of carboxymethyl-pyridinium chloride with DMAD [13].

Reaction of the pyridinium bromide 2a with pyrazole-5-diazonium chloride (7) in ethanol under neutral conditions afforded the corresponding hydrazonoyl bromide 8a in a high yield. Treatment of the latter salt with sodium carbonate in aqueous ethanol at room temperature furnished a browncolored product that was elucidated as 3-(2-benzothiazolylcarbonyl)-1-(3-phenyl-1*H*-pyrazol-5-yl)-8a*H*-1,2,4-triazolo[4,3-*a*]pyridine (**10a**), Scheme 1. The <sup>1</sup>H NMR spectrum of the latter product exhibited three singlet signals at  $\delta$  5.90, 6.61, and 9.44 (D<sub>2</sub>O-exchangeable) due to triazoline-5-CH, pyrazole-4-CH, and pyrazole-NH protons, respectively, in addition to aromatic multiplets in the region  $\delta$  7.35–8.56. Similar treatment of the pyridinium bromide salt 2b with pyrazole-5-diazonium chloride 7 under the same reaction conditions resulted in the formation of the corresponding hydrazonoyl pyridinium bromide salt 8b in 85% yield. Treatment of the latter salt 8b with sodium carbonate was performed similar to 8a, to afford the corresponding 8aH-1,2,4-triazolo [4,3-a] pyridine derivative **10b**, as outlined in Scheme 1. Formation of compounds **10a,b** from **8a,b** is assumed to take place through the elimination of the HBr molecule from **8a,b** to give the nonisolable nitrogen ylide **9a,b**, which undergoes an intramolecular 1,5-dipolar cycloaddition to give the triazolo[4,3-a]pyridine derivatives **10a,b**.

The reactivity of 1-(1-methylbenzimidazol-2-vl)-1-ethanone-2-dimethylsulfonium bromide (11) towards the synthesis of hydroximovl and hydrazonovl halides was also explored. Thus, treatment of the sulfonium bromide 11 with sodium nitrite in a mixture of dioxane/water in the presence of hydrochloric acid at room temperature afforded a good yield of a greenish-yellow-colored product that was identified as 2-(1-methylbenzimidazolyl) carbonylhydroximoyl chloride (12) (Scheme 2) on the basis of its elemental and spectral analyses. Compound 12 reacted with 2-methylthio-1*H*benzimidazole (13) in ethanol/triethylamine at refluxing conditions to give a pale yellow colored product established as 3-[(1-methylbenzimidazol-2-yl)carbonyl]benzimidazo-[1,2-d]-1,2,4-oxadiazole (14) (Scheme 2) on the basis of the elemental and spectral analyses of the reaction product. The IR spectrum of compound 12 showed a carbonyl absorption at 1693 cm<sup>-1</sup> and a broad OH band at 3350-3100 cm<sup>-1</sup>; however, the IR spectrum of 14 showed only a carbonyl absorption at 1720 cm<sup>-1</sup>.

Similarly, the hydroximoyl chloride **12** reacted with 5-amino-3-phenylpyrazole (**15**) in ethanolic triethylamine solution, at refluxing temperature, and furnished only one product as examined by TLC. The structure of the obtained product was substantiated from its elemental analysis and spectral data and identified as 3-[(1-methylbenzimidazol-2-yl)carbonyl]-6-phenylpyrazolo[1,5-c]-1,2,4-triazole (**16**), as depicted in Scheme 2. The IR spectrum of compound **16** showed C=O and NH absorption peaks at 1654 and 3153 cm<sup>-1</sup>, respectively.

It is well known that sulfonium bromides couple easily with diazonium salts in the presence of sodium acetate as a basic medium to give the corresponding hydrazonovl bromides [14]; however, in this work we could perform the coupling of the sulfonium bromide 11 with pyrazole-5-diazonium chloride (7) in ethanol under neutral condition to obtain the hith*erto* unreported α-oxo-N-(3-phenyl-1H-pyrazol-5yl)-2-(1-methyl-benzimidazole)ethane-hydrazonoyl bromide (17), having heterocyclic rings at both Cand N-terminals, in a good yield (Scheme 2). The IR spectrum of the latter hydrazonoyl bromide revealed one carbonyl and two NH absorption bands at 1654, 3210, and 3146 cm<sup>-1</sup>, respectively. Treatment of 17 with triethylamine in refluxing ethanol resulted in the formation of a product that was found to be identical in all respects (mp, mixed mp, and

spectral data) with compound 16 that was obtained above from the reaction of hydroximoyl chloride 12 with 5-amino-3-phenylpyrazole (15) as shown in Scheme 2.

#### **EXPERIMENTAL**

Melting points were measured with a Gallenkamp apparatus and are uncorrected. IR spectra were recorded on Shimadzu FT-IR 8101 PC infrared spectrophotometer. The <sup>1</sup>H NMR spectra were determined in CDCl<sub>3</sub> or DMSO-d<sub>6</sub> at 300 MHz on a Varian Mercury VX 300 NMR spectrometer using TMS as an internal standard. Mass spectra were measured on a GCMS-QP1000 EX spectrometer at 70 eV. Elemental analyses were carried out at the Microanalytical Center of Cairo University. Pyridinium salts **2a,b** [10], 5aminopyrazole derivatives 7 and 10 [15], sulfonium salts **11a** [16] and **11b** [10], and 2-methylthio-1*H*benzimidazole (13) [17] were prepared according to the procedures reported in the literature.

## *Indolizine Derivatives* **6a,b**

To a mixture of **2a** or **2b** (2 mmol) and dimethyl acetylenedicarboxylate (DMAD) (0.57 g, 4 mmol) in dry benzene (30 mL), triethylamine (0.4 mL) was added and the reaction mixture was refluxed for 3 h, and then left to cool to room temperature. The triethylamine-hydrobromide salt was removed by filtration, and the filtrate was evaporated under vacuum. The residue was triturated with methanol, where a yellow-colored precipitate was formed that was filtered off, washed with methanol, and dried. Recrystallization from DMF/EtOH afforded **6a,b**.

**6a**: Yield 79%; mp 170–172°C; IR (KBr) ν 1732, 1695 (2 C=O), 1610 (C=C) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO $d_6$ )  $\delta$  3.56 (s, 3H, CH<sub>3</sub>), 3.85 (s, 3H, CH<sub>3</sub>), 7.36 (td, 1H, J = 7.2, 1.2 Hz); 7.61–7.72 (m, 3H), 8.12 (dd, 1H, J = 7.2, 2.4 Hz), 8.26 (dd, 1H, J = 6.9, 2.4 Hz), 8.30 (d, 1H, J = 8.7 Hz), 9.54 (d, 1H, J = 6.9 Hz); <sup>13</sup>C NMR (DMSO) δ 52.47, 53.06, 105.68, 117.86, 119.51, 120.03, 123.66, 125.41, 128.13, 128.43, 129.47, 130.47, 133.12, 136.72, 138.65, 153.13, 163.19, 165.34, 167.37, 175.99; MS m/z, 394 (M<sup>+</sup>), 344, 285, 256, 160, 128, 96, 64. For C<sub>20</sub>H<sub>14</sub>N<sub>2</sub>O<sub>5</sub>S Calcd: C, 60.91; H, 3.58; N, 7.10; S, 8.13%. Found: C, 61.14; H, 3.44; N, 7.21; S, 8.19%.

**6b**: Yield 83%; mp 238–240°C; IR (KBr) ν 1720, 1713 (2 C=O), 1608 (C=C) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO $d_6$ )  $\delta$  3.64 (s, 3H, CH<sub>3</sub>), 3.91 (s, 3H, CH<sub>3</sub>), 4.26 (s, 3H, CH<sub>3</sub>), 7.28 (m, 1H); 7.54–7.68 (m, 5H), 8.05 (d, 1H, J = 8.6 Hz), 8.63 (d, 1H, J = 7.2 Hz); MS m/z,  $392 (M^+ + 1)$ ,  $391 (M^+)$ , 362, 332, 300, 246, 165, 143, 116, 77. For C<sub>21</sub>H<sub>17</sub>N<sub>3</sub>O<sub>5</sub> Calcd: C, 64.45; H, 4.38; N, 10.74%. Found: C, 64.36; H, 4.21; N, 10.44%.

# Hydrazonovl Pyridinium Bromides 8a,b

A mixture of **2a** or **2b** (5 mmol) and 3-phenylpyrazole-5-diazonium chloride (7) (1.03 g, 5 mmol) in absolute ethanol (50 mL) was left to stir at room temperature for 8 h. The orange-yellow-colored precipitated products were filtered off, washed with absolute ethanol, and dried. Recrystallization from acetic acid afforded **8a.b**.

**8a**: Yield 73%; mp  $>300^{\circ}$ C; IR (KBr)  $\nu$  3404 (br 2NH), 1714 (C=O), 1629 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  6.49 (s, 1H), 7.36–7.80 (m, 7H); 8.16–  $8.23 \, (m, 2H), 8.46 - 8.61 \, (m, 3H), 9.32 \, (d, 2H, J = 6.27)$ Hz), 10.17 (br s, 1H, NH), 13.12 (br s, 1H, NH). For C<sub>23</sub>H<sub>17</sub>BrN<sub>6</sub>OS Calcd: C, 54.66; H, 3.39; N, 16.63; S, 6.34%. Found: C, 54.58; H, 3.37; N, 16.44; S, 6.38%.

**8b**: Yield 80%; mp >300°C; IR (KBr)  $\nu$  3226, 3132 (2NH), 1697 (C=O), 1616 (C=N) cm<sup>-1</sup>. For C<sub>24</sub>H<sub>20</sub>BrN<sub>7</sub>O Calcd: C, 57.38; H, 4.01; N, 19.52%. Found: C, 57.50; H, 3.77; N, 19.81%.

# 1,2,4-Triazolo[4,3-a]pyridine **10a,b**

To a stirred solution of **8a** or **8b** (2 mmol) in ethanol (20 mL) and water (10 mL) was added sodium carbonate solution (0.3 g in 5-mL water) portionwise. The reaction mixture was left to stir at room temperature for 4 h. The brown-colored precipitate was filtered off, washed with water and then ethanol, and dried. Recrystallization from DMF afforded 10a,b.

**10a**: Yield 66%; mp 238–240°C; IR (KBr) ν 3210 (NH), 1665 (C=O), 1610 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  5.90 (s, 1H, triazoline-5-CH), 6.61 (s, 1H, pyrazole-4-CH), 7.35–7.83 (m, 6H); 8.05–8.20 (m, 3H), 8.33-8.56 (m, 4H), 9.44 (br s, 1H, NH). MS m/z,  $425 (M^+ + 1), 424 (M^+), 345, 292, 212, 161, 135, 81,$ 77. For C<sub>23</sub>H<sub>16</sub>N<sub>6</sub>OS Calcd: C, 65.08; H, 3.80; N, 19.80; S, 7.55%. Found: C, 65.21; H, 3.84; N, 19.59; S, 7.47%.

**10b**: Yield 58%; mp 262–264°C; IR (KBr) ν 3192 (NH), 1670 (C=O), 1612 (C=N) cm<sup>-1</sup>; MS m/z, 422  $(M^+ + 1)$ , 421  $(M^+)$ , 319, 290, 195, 128, 105, 79. For C<sub>24</sub>H<sub>19</sub>N<sub>7</sub>O Calcd: C, 68.40; H, 4.54; N, 23.26%. Found: C, 68.21; H, 4.84; N, 23.55%.

## Hydroximoyl Chloride 12

To a stirred mixture of 11 (3.15 g, 10 mmol) and sodium nitrite (0.7 g, 10 mmol) in dioxane/ water (40 mL, 1:1) was added conc. HCl (50 mL) portionwise over a period of 1 h. The reaction mixture was left to stir for further 2 h at room temperature. The greenish-yellow-colored precipitate was filtered off, washed with water, and dried. Recrystallization from toluene afforded 12 in 63% yield; mp 178–180°C; IR (KBr) ν 3350–3100 (br OH), 1693

(C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  4.22 (s, 3H), 7.44–7.61 (m, 2H); 8.05 (d, 1H, J = 7.8 Hz), 8.17 (d, 1H, J = 8.1 Hz), 9.1 (s, 1H). For C<sub>10</sub>H<sub>8</sub>ClN<sub>3</sub>O<sub>2</sub> Calcd: C, 50.54; H, 3.39; N, 17.68%. Found: C, 50.48; H, 3.22; N, 17.45%.

# Benzimidazo[1,2-d]-1,2,4-oxadiazole Derivative **14**

To a mixture of **12** (0.474 g, 2 mmol) and 2-methylthio-1*H*-benzimidazole (**13**) (0.33 g, 2 mmol) in ethanol (20 mL), triethylamine (0.2 mL) was added and the reaction mixture was refluxed for 6 h, then left to cool to room temperature. The reaction mixture was diluted with water and the so-formed precipitate was filtered off, washed with ethanol, and dried. Recrystallization from DMF/EtOH afforded **14** in 45% yield; mp 190–192°C; IR (KBr)  $\nu$  1720 (C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  4.21 (s, 3H), 7.28–7.65 (m, 6H); 8.10–8.21 (m, 2H); MS m/z, 317 (M<sup>+</sup>), 282, 210, 131, 103, 76. For C<sub>17</sub>H<sub>11</sub>N<sub>5</sub>O<sub>2</sub> Calcd: C, 64.35; H, 3.49; N, 22.07%. Found: C, 64.23; H, 3.38; N, 22.31%.

# Hydrazonoyl Bromide 17

A mixture of **11** (3.15 g, 10 mmol) and 3-phenylpyrazole-5-diazonium chloride (**7**) (2.06 g, 10 mmol) in absolute ethanol (20 mL) was left to stir at room temperature for 6 h. The orange-yellow-colored precipitate was filtered off, washed with water, and dried. Recrystallization from acetic acid afforded **17** in 59% yield; mp 205–207°C; IR (KBr)  $\nu$  3210, 3146 (2NH), 1654 (C=O), 1610 (C=N) cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.1 (s, 3H), 6.66 (s, 1H), 7.12–8.01 (m, 9H); 9.48 (br s, 1H, NH), 13.95 (br s, 1H, NH). For C<sub>19</sub>H<sub>15</sub>BrN<sub>6</sub>O Calcd: C, 53.91; H, 3.57; N, 19.86%. Found: C, 53.69; H, 3.24; N, 19.72%.

# 3-[(1-Methylbenzimidazol-2-yl)carbonyl]-6-phenylpyrazolo[1,5-c]-1,2,4-triazole (**16**)

Method A. To a mixture of **12** (0.474 g, 2 mmol) and 5-amino-3-phenylpyrazole (**15**) (0.312 g, 2 mmol) in ethanol (20 mL), triethylamine (0.2 mL) was added and the reaction mixture was refluxed for 6 h, then left to cool to room temperature. The precipitated product was filtered off, washed with water and ethanol, dried, and finally recrystallized from EtOH/DMF to afford **16** in 68% yield; mp 216–218°C; IR (KBr) ν 3153 (NH), 1654 (C=O) cm<sup>-1</sup>;

<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.32 (s, 3H), 7.26–7.31 (m, 3H), 7.4–7.51 (m, 2H), 7.66–7.76 (m, 5H), 8.87 (s, 1H); MS m/z, 343 (M<sup>+</sup> + 1), 342 (M<sup>+</sup>), 304, 276, 170, 155, 77. For C<sub>19</sub>H<sub>14</sub>N<sub>6</sub>O Calcd: C, 66.66; H, 4.12; N, 24.55%. Found: C, 66.58; H, 4.27; N, 24.38%.

Method B. To a solution of 17 (0.423 g, 1 mmol) in ethanol (20 mL), triethylamine (0.2 mL) was added and the reaction mixture was refluxed for 2 h, then left to cool to room temperature. The brown-colored precipitated product was filtered off, washed with water and ethanol, dried, and finally recrystal-lized from EtOH/DMF to afford a product identical in all respects (mp, mixed mp, and spectra) with the pyrazolo[1,5-c]-1,2,4-triazole derivative 16 in 74% yield.

## REFERENCES

- [1] Milet, R.; Domarkas; J.; Rigo, B.; Goossens, L.; Houssin, R.; Henichart, J. Bioorg Med Chem 2002, 10, 2905.
- [2] Sonnet, P.; Dallemagne, P.; Guillon, J.; Enguehard, C.; Steibing, S.; Tanguy, J.; Bureau, R.; Rault, S.; Sara, G. Bioorg Med Chem 2000, 8, 945.
- [3] Gubin, J.; Lucchetti, J.; Mahaux, J.; Nisato, D.; Rossee, G. J Med Chem 1992, 35, 981.
- [4] Flitsch, W. In Comprehensive Heterocylic Chemistry; Katrizky, A. R.; Rees, C. W.; (Eds.); Pergamon: Oxford 1984; Vol. 4, p. 476.
- [5] Weidner, C. H.; Hadsworth, D. H.; Bender, S. L.; Beltman, D. J. J Org Chem 1989, 54, 3660.
- [6] Paramashivappa, R.; Kumar, P. P.; Rao, P. V.; Rao, A. S. Bioorg Med Chem Lett 2003, 13, 657.
- [7] Roth, T.; Morningstar, M. L.; Boyer, B. L.; Hughes, S. H.; Bukheit, R. W.; Michejda, C. J. J Med Chem 1997, 40, 4199.
- [8] Soderlind, K. J.; Gorodetsky, B.; Singh, A. K.; Bachur, N. R.; Miller, G. G.; Lown, J. W.; Anti-Cancer Drug Des 1999, 14, 19.
- [9] Dawood, K. M.; Ragab, E. A.; Farag, A. M. J Chem Res 2003, (S) 685, (M) 1151.
- [10] Dawood, K. M. Synth Commun 2001, 31, 1647.
- [11] Dawood, K. M.; Kandeel, Z. E.; Farag, A. M. Heteroatom Chem 1999, 10, 417.
- [12] Dawood, K. M. J Chem Res (S) 1998, 128.
- [13] Zhang, L.; Liang, F.; Sun, L.; Hu, Y.; Hu, H. Synthesis 2002, 1733.
- [14] Shawali, A. S.; Abdallah, M. A. Adv Heterocycl Chem 1995, 63, 277.
- [15] Elnagdi, M. H.; Elmoghayar, M. R. H.; Fleita, E. A. A.; Fahmy, S. M. J Org Chem 1976, 41, 3781.
- [16] Farag, A. M.; Dawood, K. M. Heteroatom Chem 1997, 8, 45.
- [17] Van Allan, J. A. J Org Chem 1956, 21, 24.